

Optimal pH requirements for different species

John Irwin, Department of Horticultural Science, University of Minnesota

pH requirements for different species vary widely. The basis for pH preferences varies. However, for many species, optimal medium pH levels vary because species differ in how well they can take up specific nutrients. Medium pH dramatically effects nutrient availability to a plant (Figure 1). As a result, as pH increases or decreases, some of the differences between nutrient uptake among species becomes obvious. These differences can occur because species have difficulty taking up nutrients such as iron or

because some species take up too much of specific nutrients such as iron or manganese. In contrast, optimal recommended pH levels may not be related to nutrient availability at all, but rather susceptibility to disease.

As mentioned above, recommended pH levels may be related to the ability or inability of a plant species to take up a specific nutrient. For instance, many of us observed for years that as pH increases, petunias will express iron deficiency first among bedding plant species. Why?, because

petunias have a hard time taking up iron. As pH increases and iron becomes less available in the media,

petunias will show the emerging iron deficiency first! Similarly, many of

Continued on page 12

Figure 1. How nutrient availability in either soil-based medium (a) or soilless medium varies with pH levels.

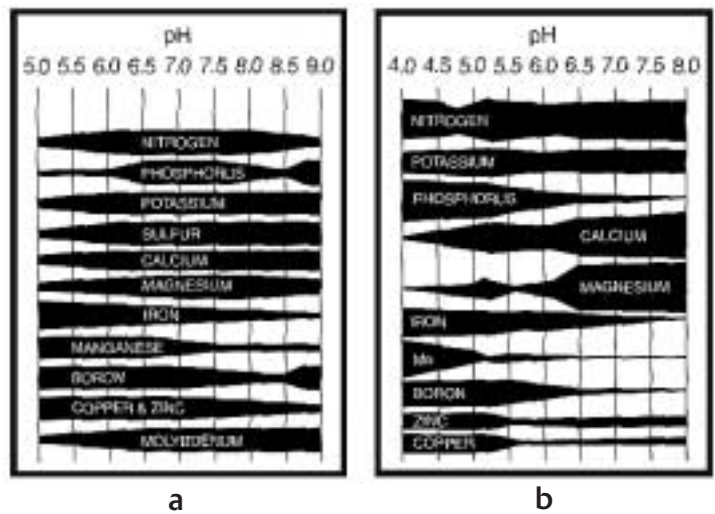


Table 1. Optimal medium pH levels for different commercial bedding plant species.

Latin Name	Common Name	Optimal pH Range	Reason
<i>Antirrhinum majus</i>	Snapdragon	5.5-6.2	Difficulty taking up iron/calcium at higher pH
<i>Browallia speciosa</i>	Browallia	5.5-6.2	--
<i>Catharanthes roseum</i>	Vinca	5.5-6.2	Difficulty taking up iron or calcium/ Thelaviopsis susceptibility at higher pH
<i>Eustoma grandiflorum</i>	Lisianthus	6.2-6.8	--
<i>Impatiens hawkeri</i>	New Guinea Impatiens	6.2-6.8	Will take up excessive amounts of Fe and Mn at low pH
<i>Primula vulgaris</i>	Primrose	5.5-6.2	Difficulty taking up micronutrients at higher pH
<i>Pelargonium x hortorum</i>	Geranium	6.2-6.8	Will take up excessive amounts of iron and manganese at low pH
<i>Petunia x hybrida</i>	Petunia	5.5-6.2	Difficulty taking up iron
<i>Salvia splendens</i>	Salvia	5.5-6.2	Difficulty taking up iron/calcium at higher pH
<i>Scaveola aemula</i>	Scaveola	5.5-6.2	Difficulty taking up iron
<i>Sutera spp</i>	Bacopa	5.5-6.2	Difficulty taking up iron
<i>Tagetes erecta</i>	African Marigold	6.2-6.8	Will take up excessive amounts of Fe and Mn at low pH
<i>Torenia fournieri</i>	Torenia	5.5-6.2	Difficulty taking up iron at higher pH
<i>Verbena x hybrida (vegetative only)</i>	Vegetative Verbena	5.5-6.2	Difficulty taking up iron
<i>Viola x wittrockiana</i>	Pansy	5.5-6.0	Susceptible to Thelaviopsis at higher pH

pH, continued from page 11

the new vegetatively propagated annuals also have pH preferences that are specific. For instance, *Scaveola* will express iron deficiency when pH levels reach 6.5.

In contrast to medium pH limiting nutrient availability, many of us learned “the hard way” that New Guinea impatiens, seed geraniums and/or

African marigolds grown at medium pH levels less than 6.0 will develop iron and/or manganese toxicity. This nutritional problem occurs because these species are very efficient in taking up iron and manganese. When medium pH levels decrease, and iron and manganese become very available, these plants take up too much iron and manganese and can literally kill themselves!

Lastly, media pH can affect susceptibility of plants to a disease or

disease growth, itself. For instance, when medium pH levels exceed 6.2-6.5 the susceptibility of plant species to *Thelaviopsis* increases dramatically. For this reason, pansies, or other species susceptible to *Thelaviopsis*, should never be grown at medium pH levels over 6.0! If pH levels exceed 6.5, consider drenching with Cleary's 3336 (8oz/100 gallons).

The table on the previous page (Table 1) shows optimal medium pH ranges for different plant

species. As shown in this table, the basis for differences in optimal pH among species differs. Along with showing which species will show which problems, each of these species is a good ‘indicator plant’ for either low or high pH problems. The only problem is, by the time you see a problem, it has been happening for awhile and crop growth has already been reduced!

